

Different coordination abilities of 1,7- and 4,7-phenanthroline in the reactions with copper(II) salts: structural characterization and biological evaluation of the reaction products

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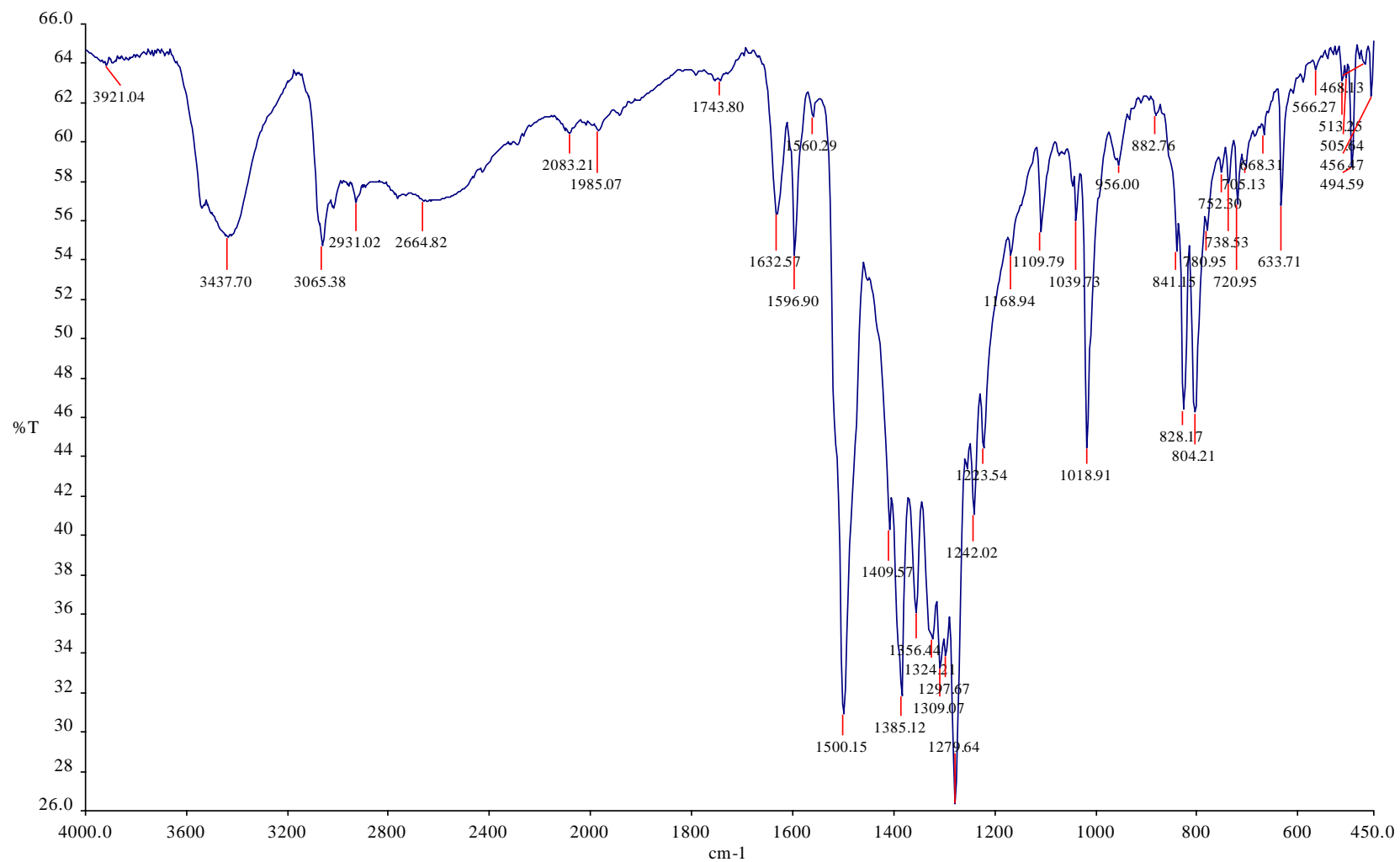
Abstract

The reactions between equimolar amounts of CuX_2 ($\text{X} = \text{NO}_3^-$ and CF_3SO_3^-) and two aromatic nitrogen-containing heterocycles differing in the position of nitrogen atoms, 1,7- and 4,7-phenanthroline (1,7- and 4,7-phen), were performed in ethanol/methanol at room temperature. When CuX_2 salts were mixed with 4,7-phen, two copper(II) complexes, $[\text{Cu}(\text{NO}_3)_2(4,7\text{-Hphen})_2](\text{NO}_3)_2$ (**1**) and $[\text{Cu}(\text{CF}_3\text{SO}_3)(4,7\text{-phen})_2(\text{H}_2\text{O})_2]\text{CF}_3\text{SO}_3$ (**2**), were formed. On the other hand, in the reaction of CuX_2 salts with 1,7-phen, only 1,7-Hphen NO_3 (**3a/b**) and 1,7-Hphen CF_3SO_3 (**4**) were obtained as the final products. The obtained products **1** – **4** were characterized by spectroscopic and X-ray diffraction techniques. In the copper(II) complexes **1** and **2**, the coordination geometry around the Cu(II) ion is distorted octahedral and square pyramidal, respectively. The antimicrobial potential of the copper(II) complexes **1** and **2** and corresponding compounds used for their synthesis were assessed against four different bacterial species and *Candida albicans*, displaying moderate growth inhibiting activity. The cytotoxic properties of the investigated complexes were also evaluated against the normal human lung fibroblast cell line (MRC-5) indicating moderate, yet more pronounced cytotoxicity than antimicrobial properties.

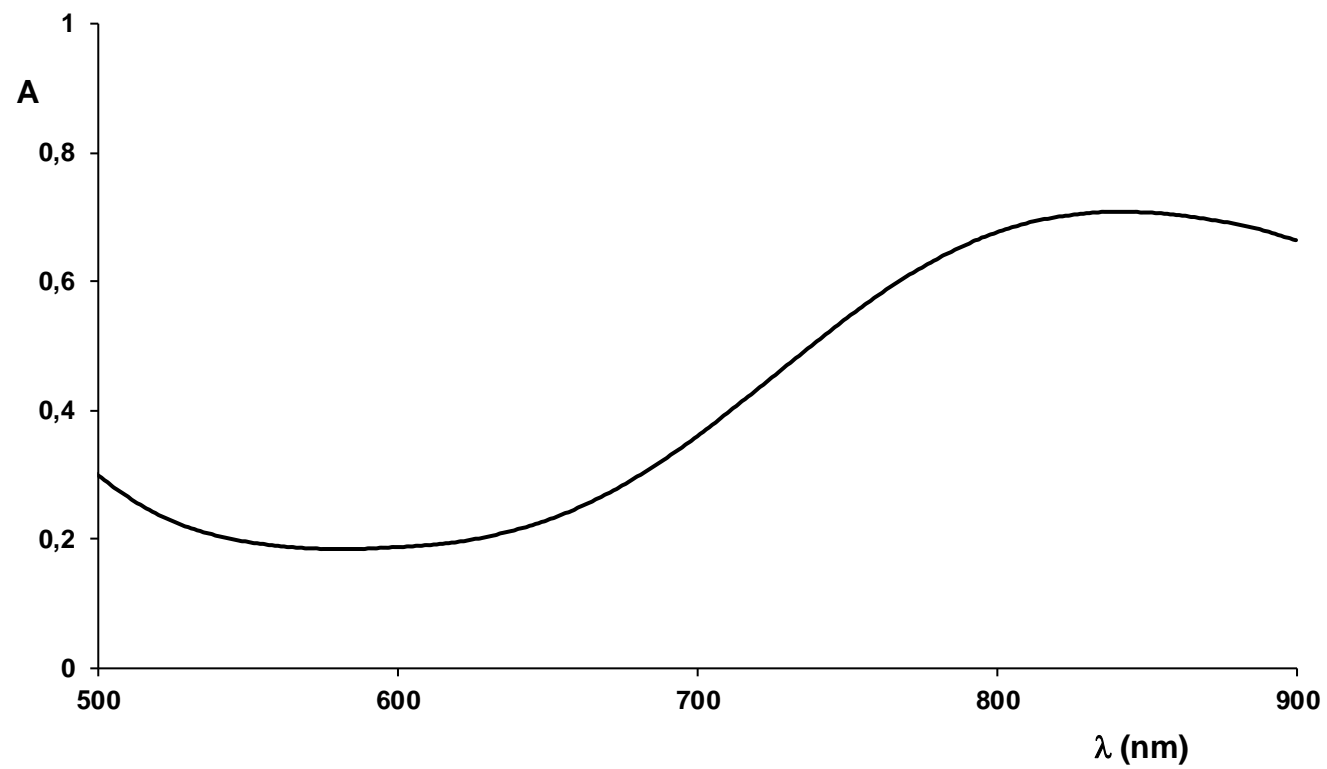
Keywords: Copper(II), Phenanthroline, Structural characterization, Antimicrobial activity, Cytotoxicity

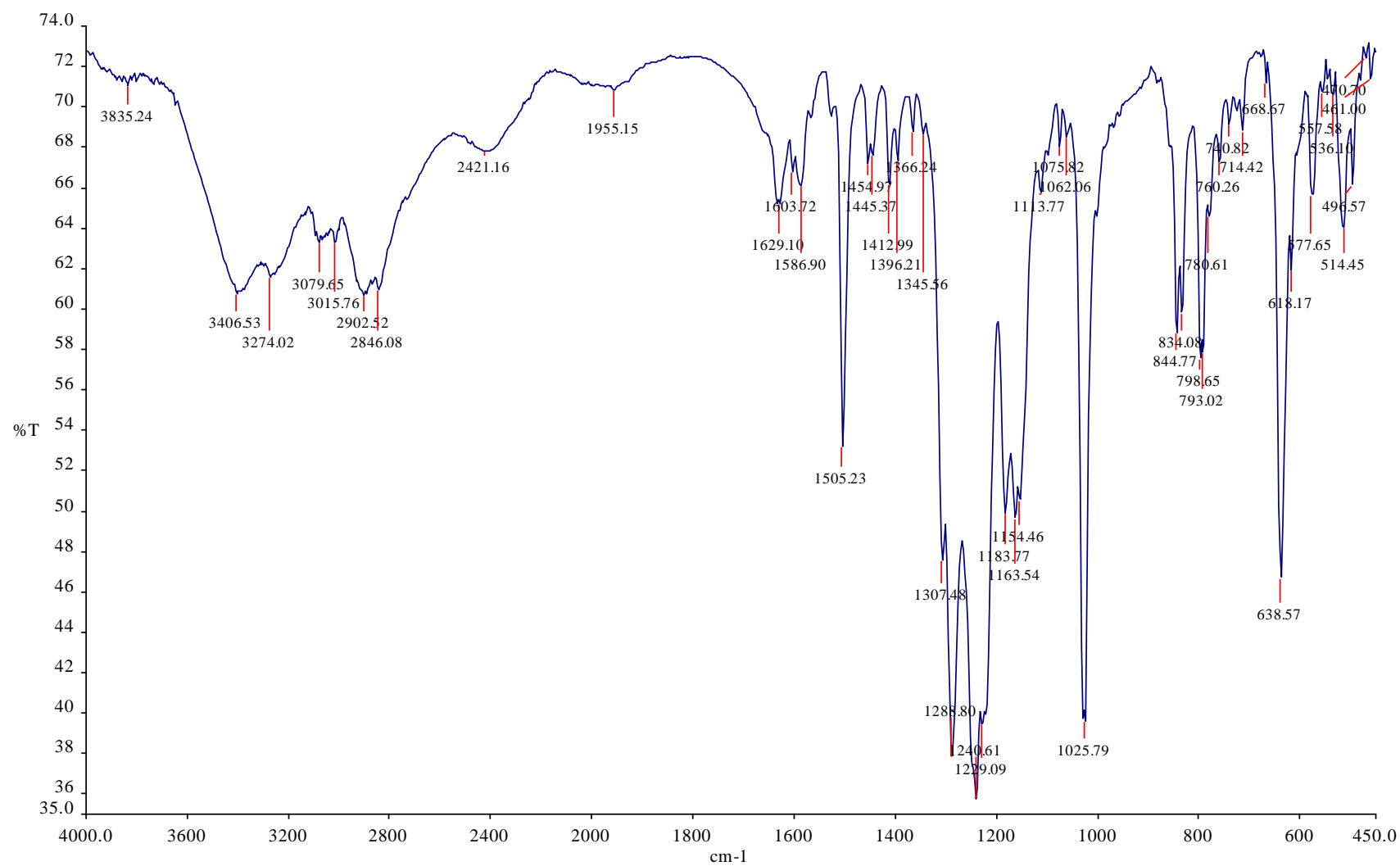
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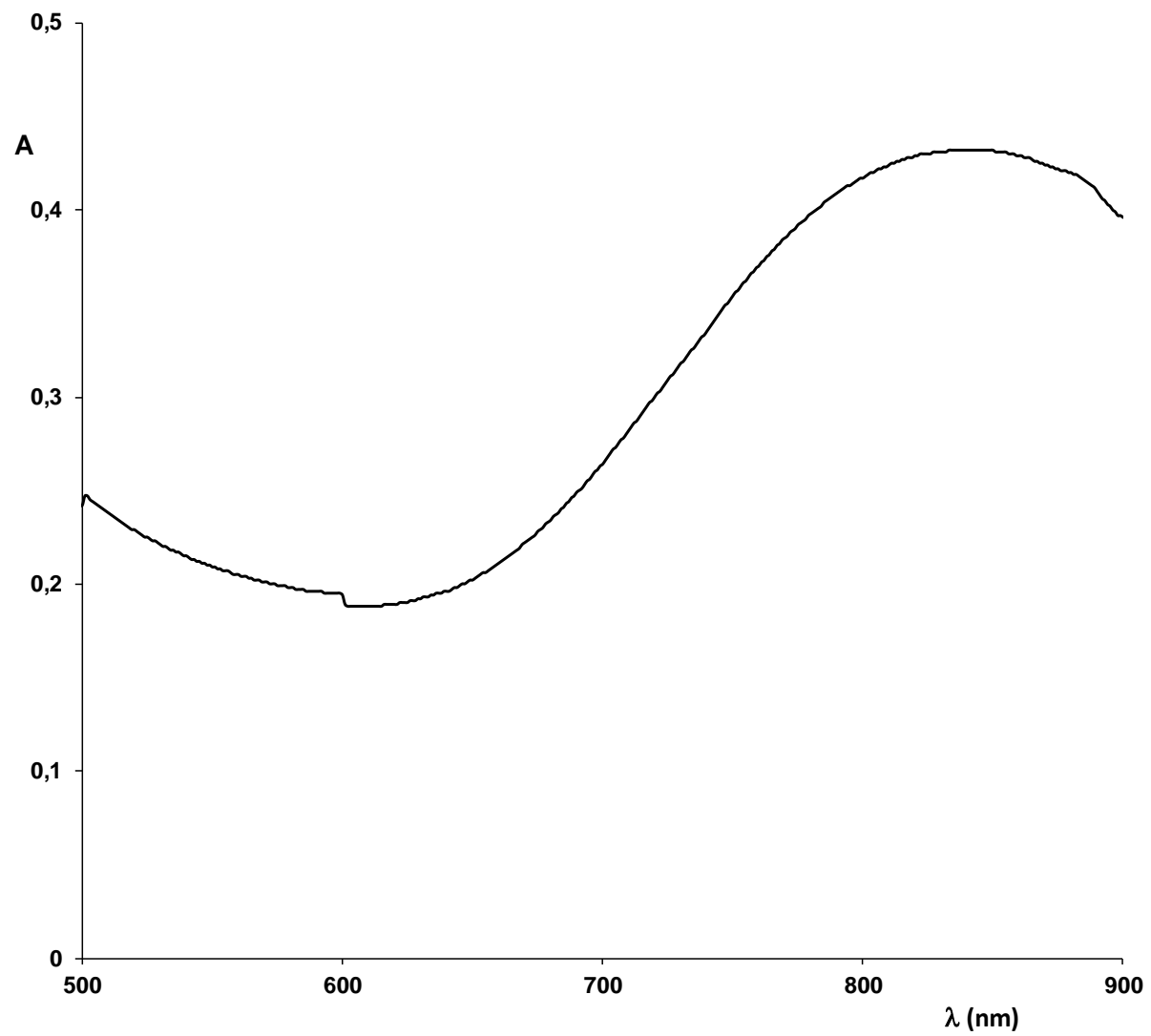
IR spectrum of **1**

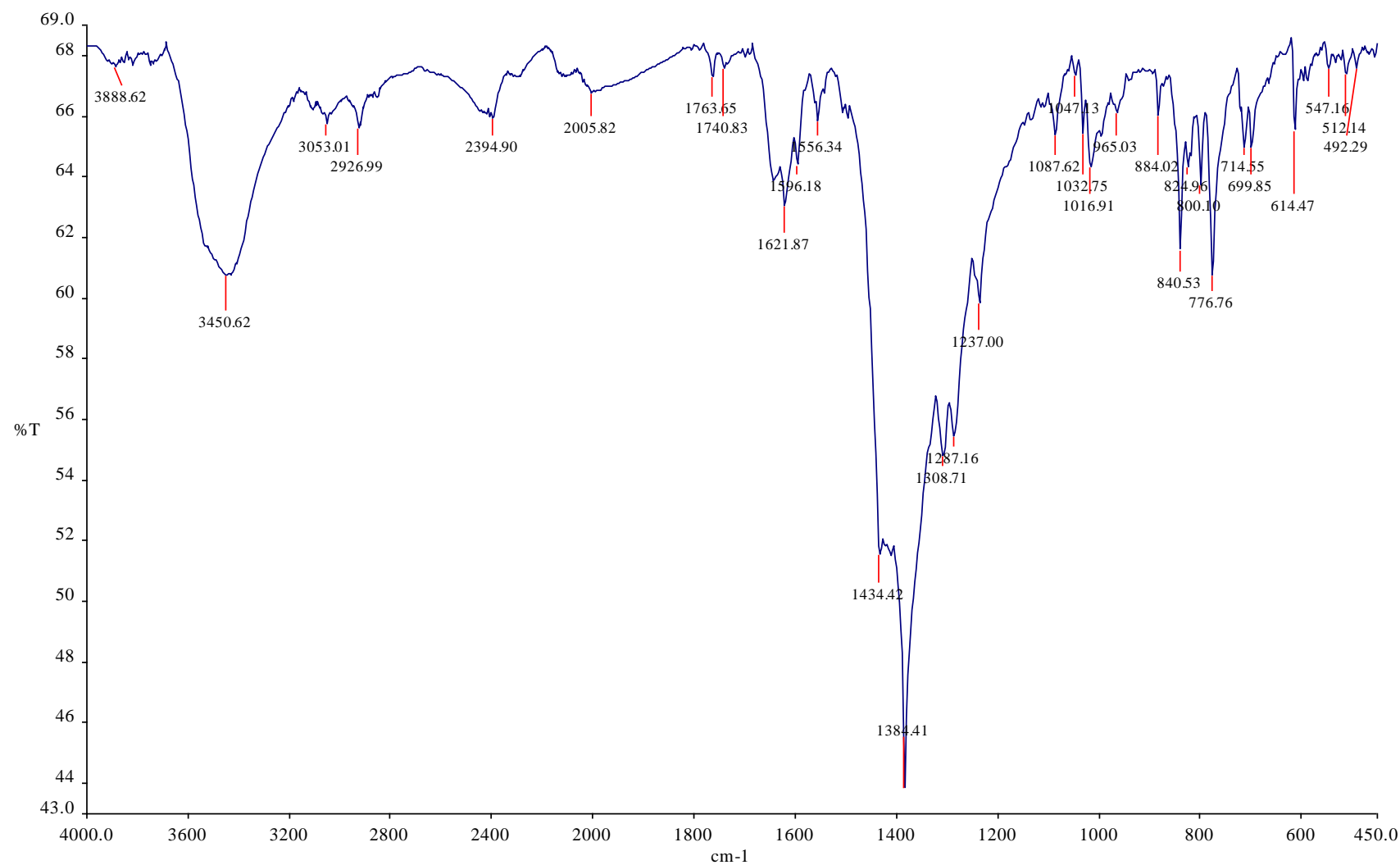
UV-Vis spectrum of **1** (DMSO)



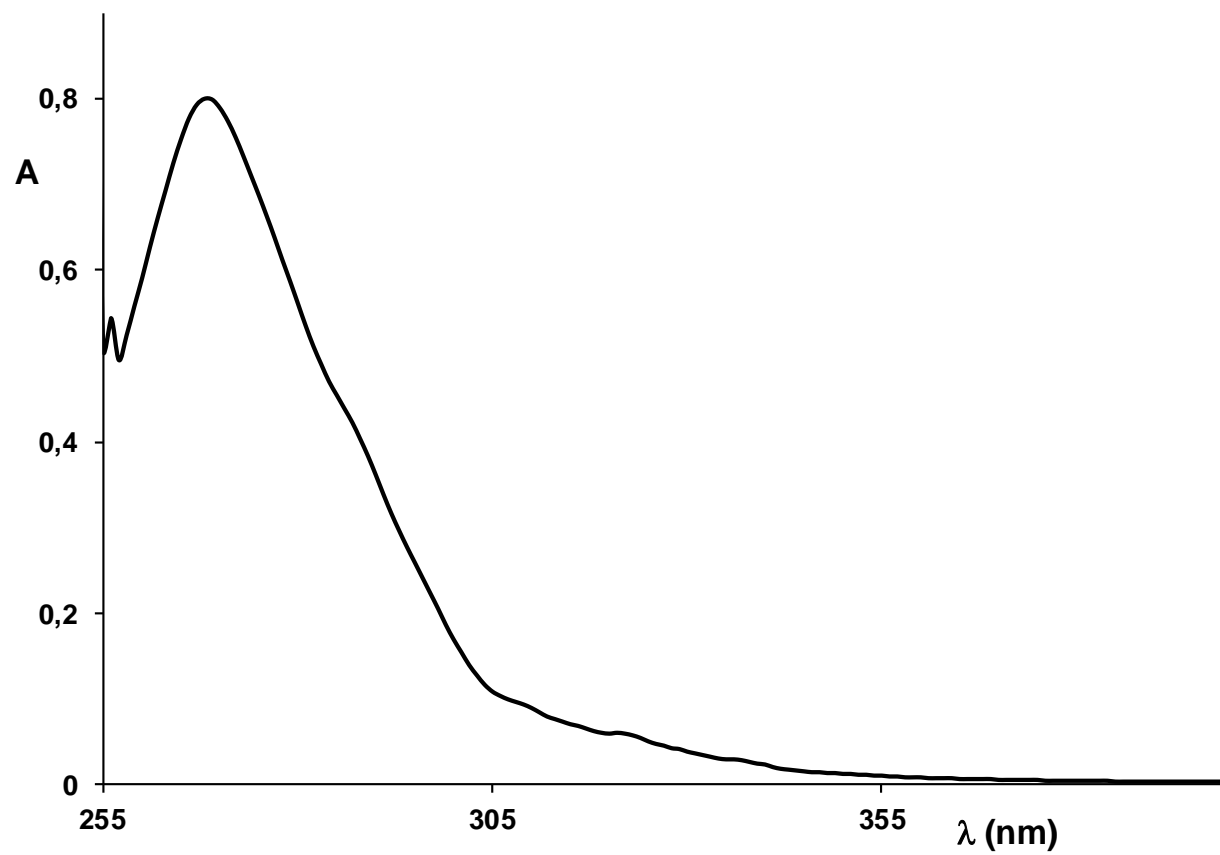
IR spectrum of **2**

UV-Vis spectrum of **2** (DMSO)

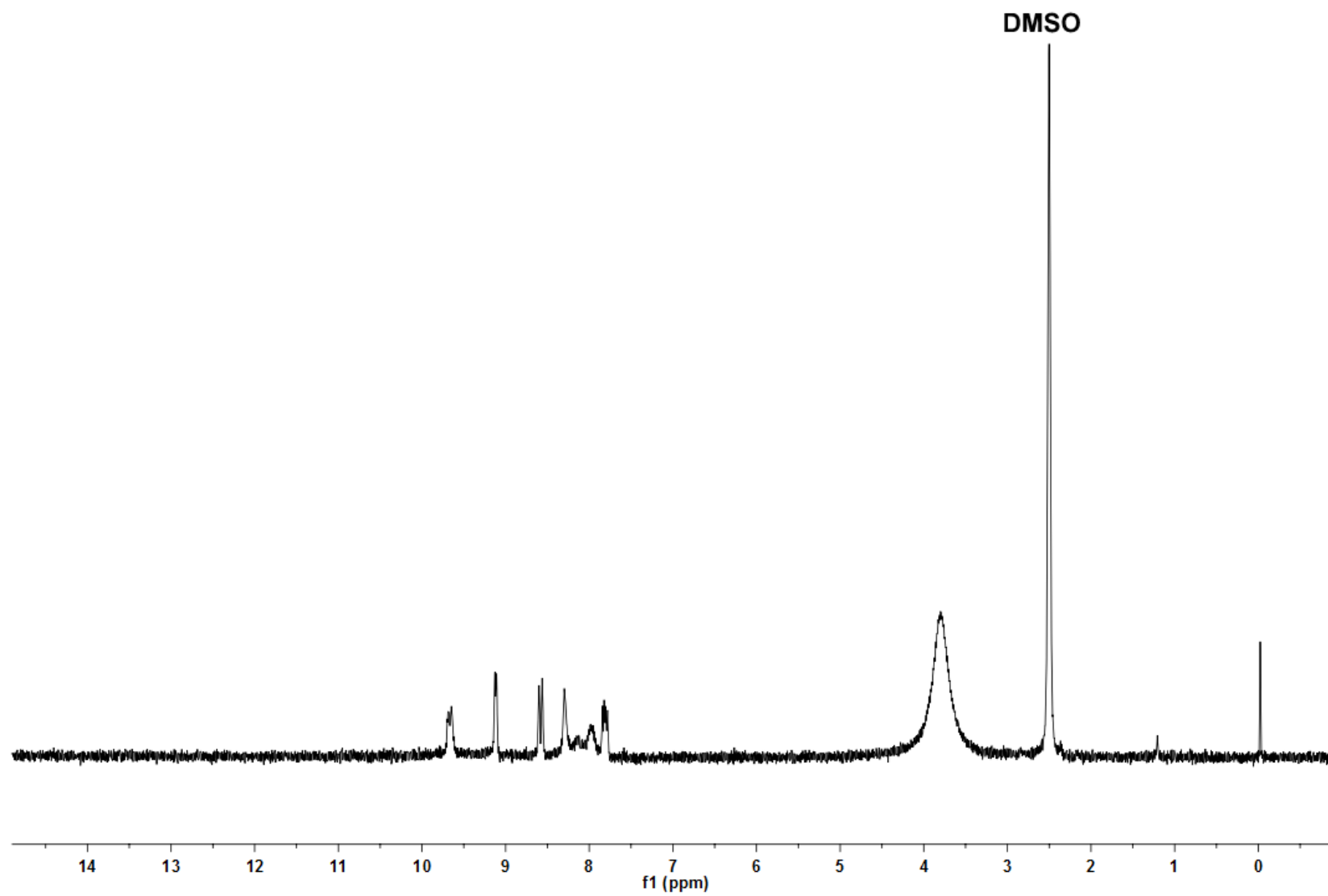


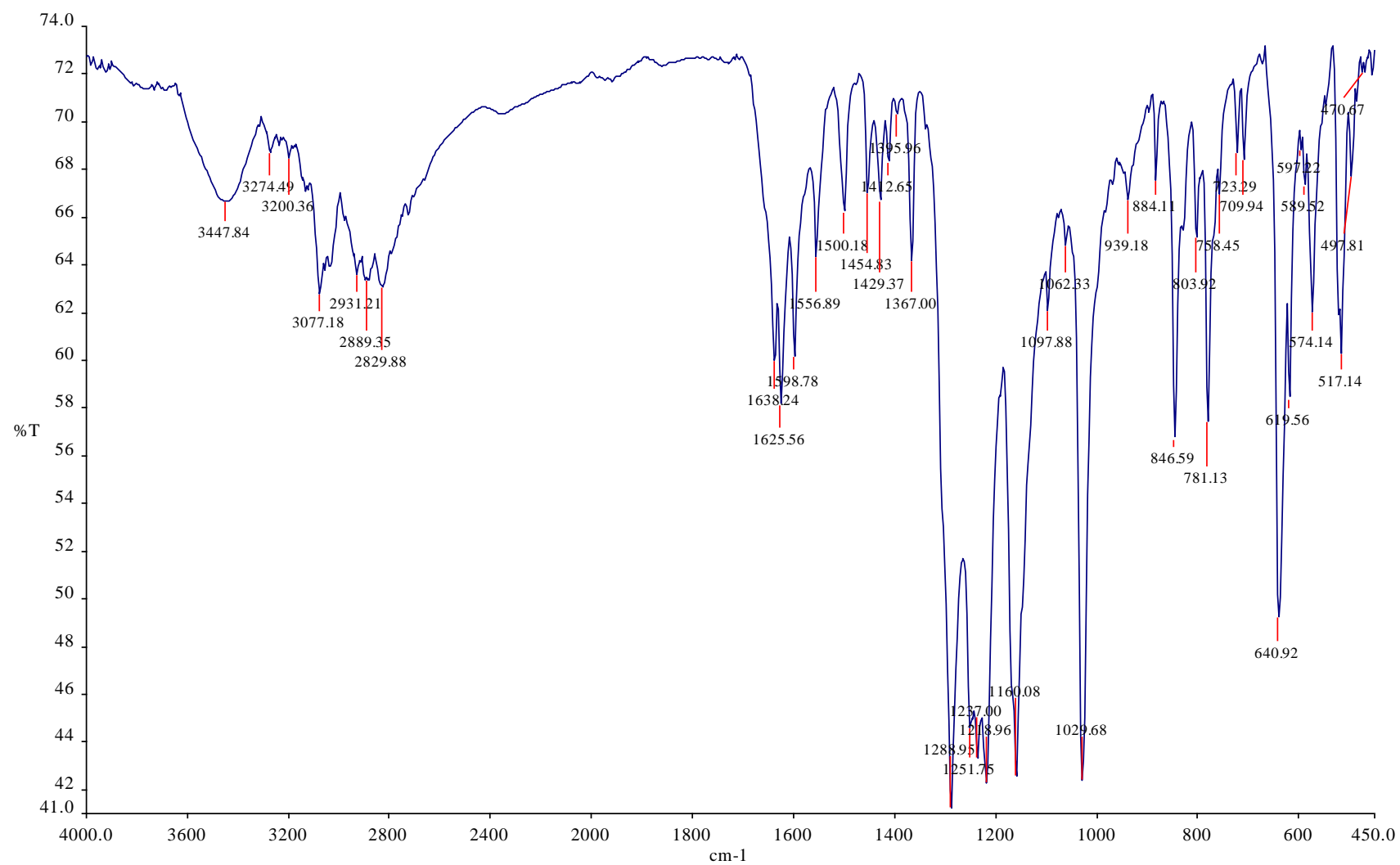
IR spectrum of **3a**

UV-Vis spectrum of **3a** (DMSO)

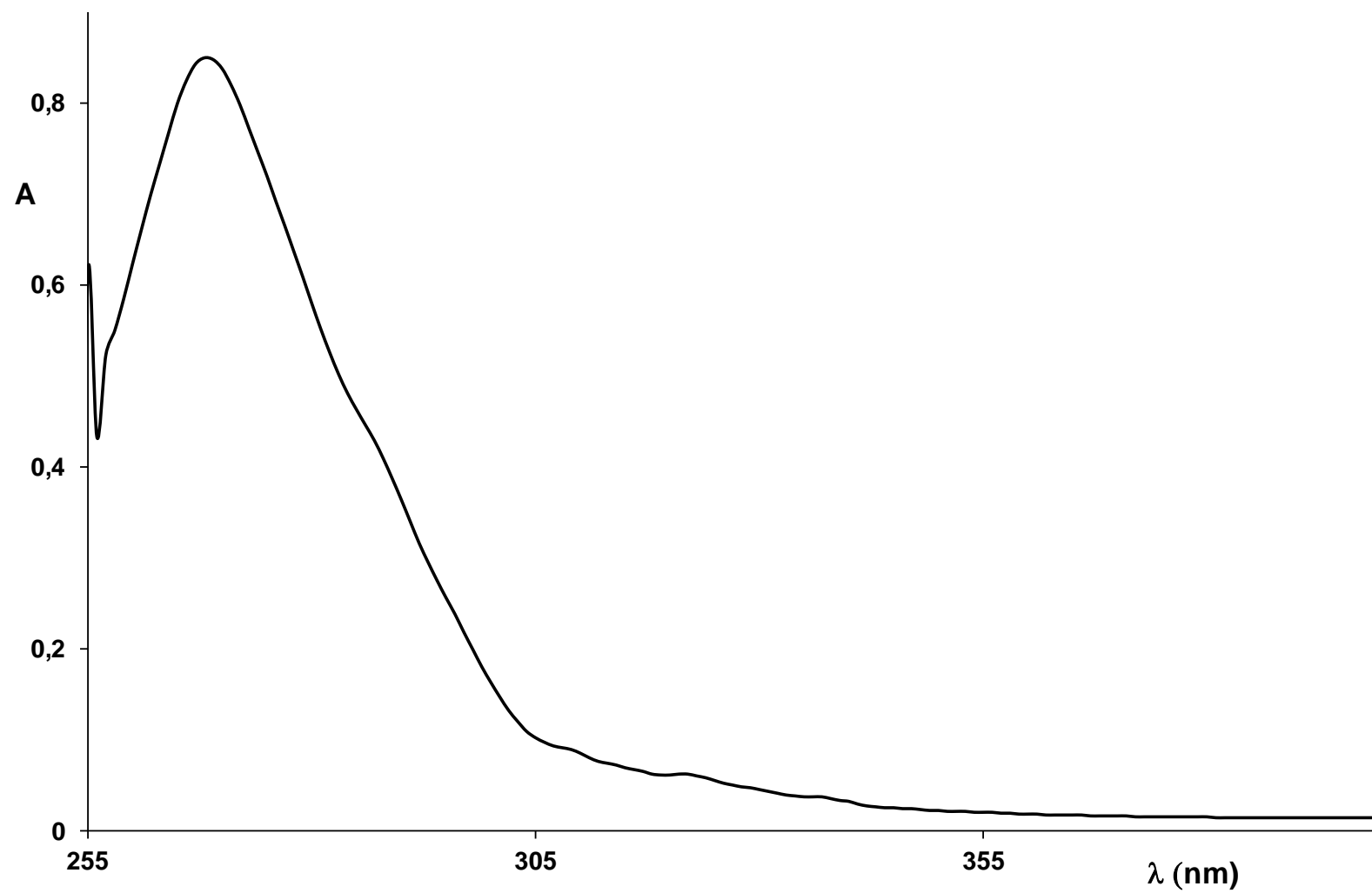


^1H NMR spectrum of **3a** (200 MHz, DMSO- d_6)

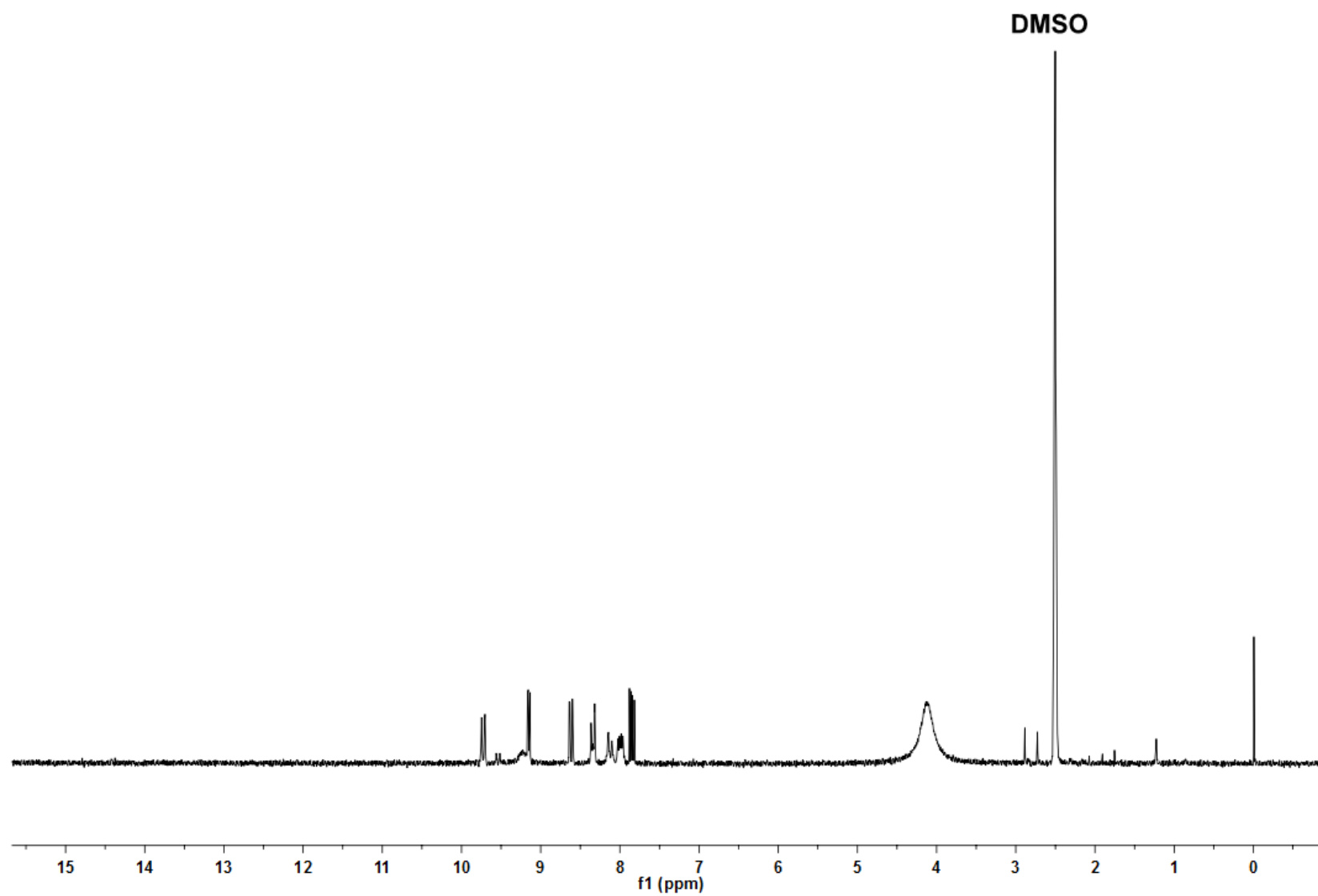


IR spectrum of **4**

UV-Vis spectrum of **4** (DMSO)



^1H NMR spectrum of **4** (200 MHz, $\text{DMSO}-d_6$)



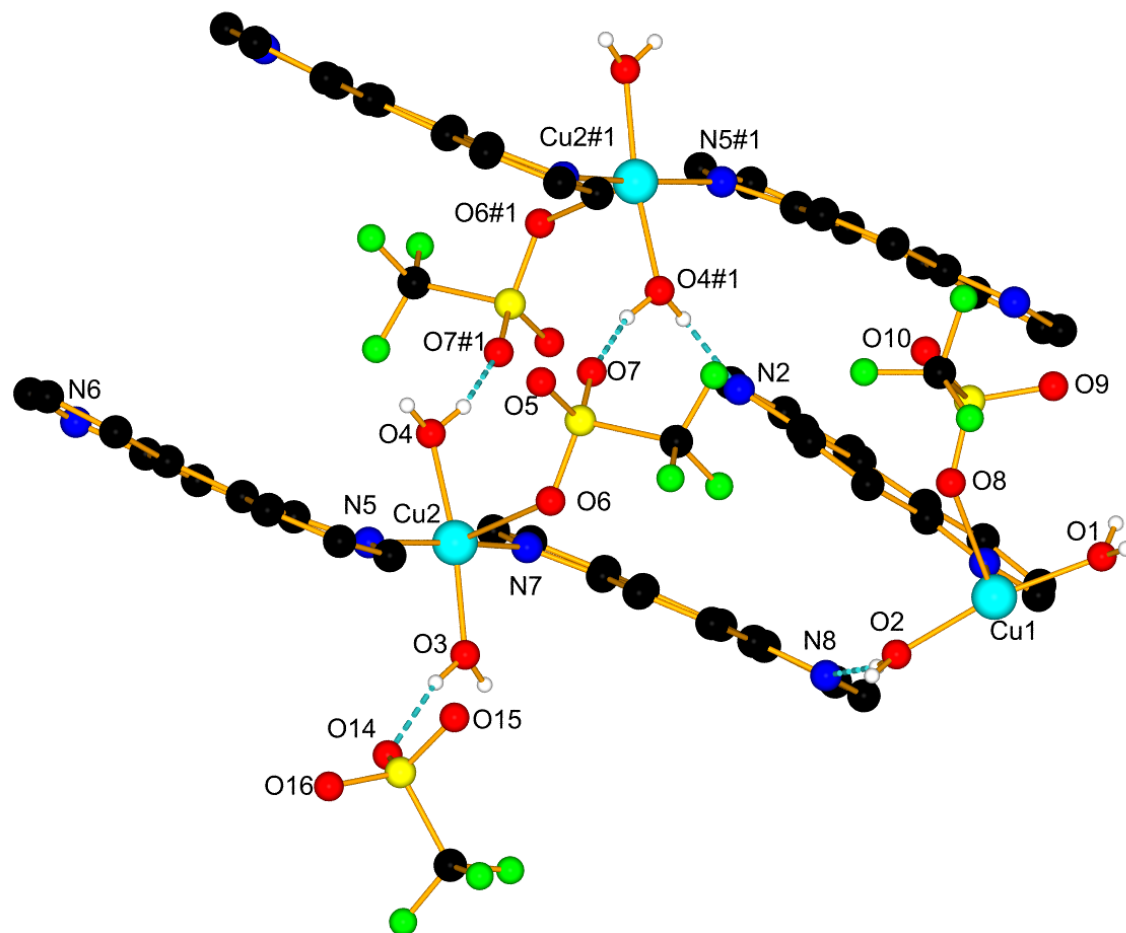


Fig. S1. Molecular structure of **2**. Some hydrogen atoms and partitioning of counter-ions are omitted for clarity. Hydrogen bonds are drawn as dashed blue lines, #1: $-x+1$, $-y+1$, $-z+1$.

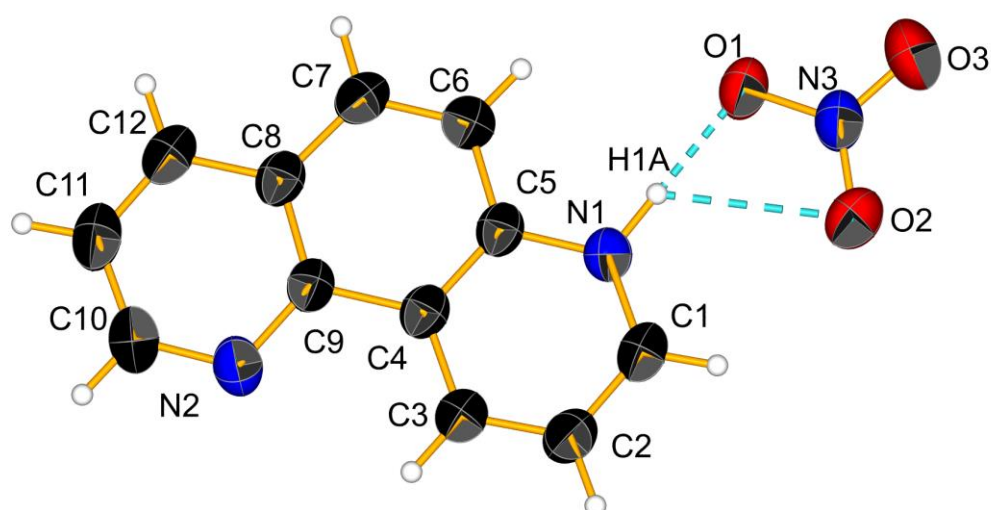


Fig. S2. Molecular structure of **3b**. Displacement ellipsoids are drawn at 50% probability level and H atoms are represented by spheres of arbitrary size.

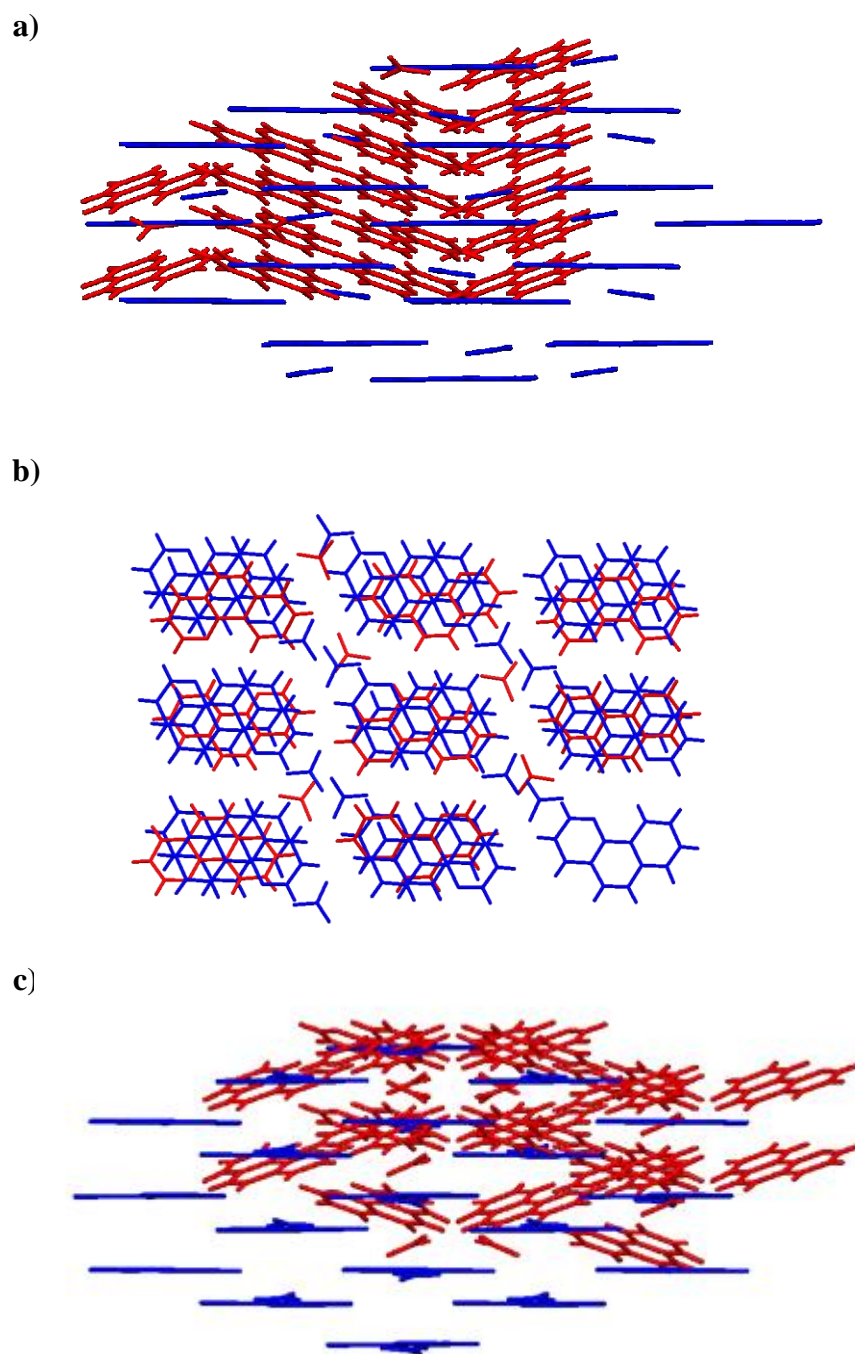


Fig. S3. The overlay of the packing of the **3a** (red) and **3b** (blue) along the *a* (**a**), *b* (**b**) and *c* (**c**) axis.