



Supporting Information

for

Tautomerism as primary signaling mechanism in metal sensing: the case of amide group

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Beilstein J. Org. Chem. **2019**, *15*, 1898–1906. [doi:10.3762/bjoc.15.185](https://doi.org/10.3762/bjoc.15.185)

Additional experimental and calculated data

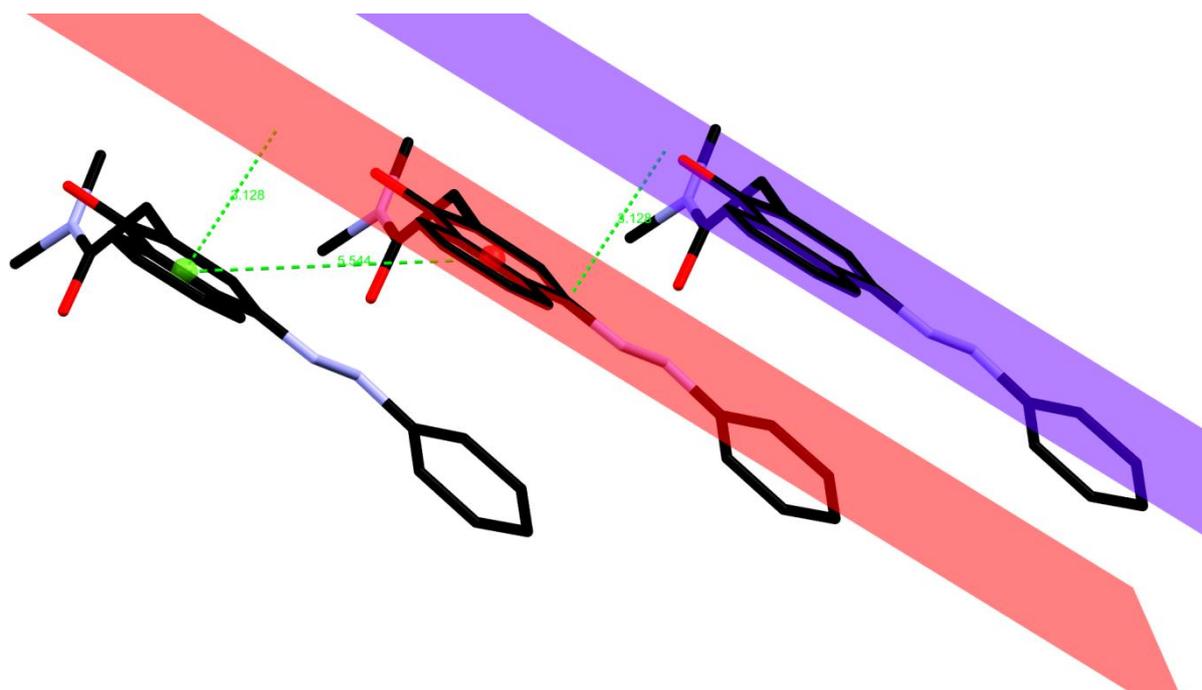


Figure S1: Distance between monomers in the aggregate crystal structure.

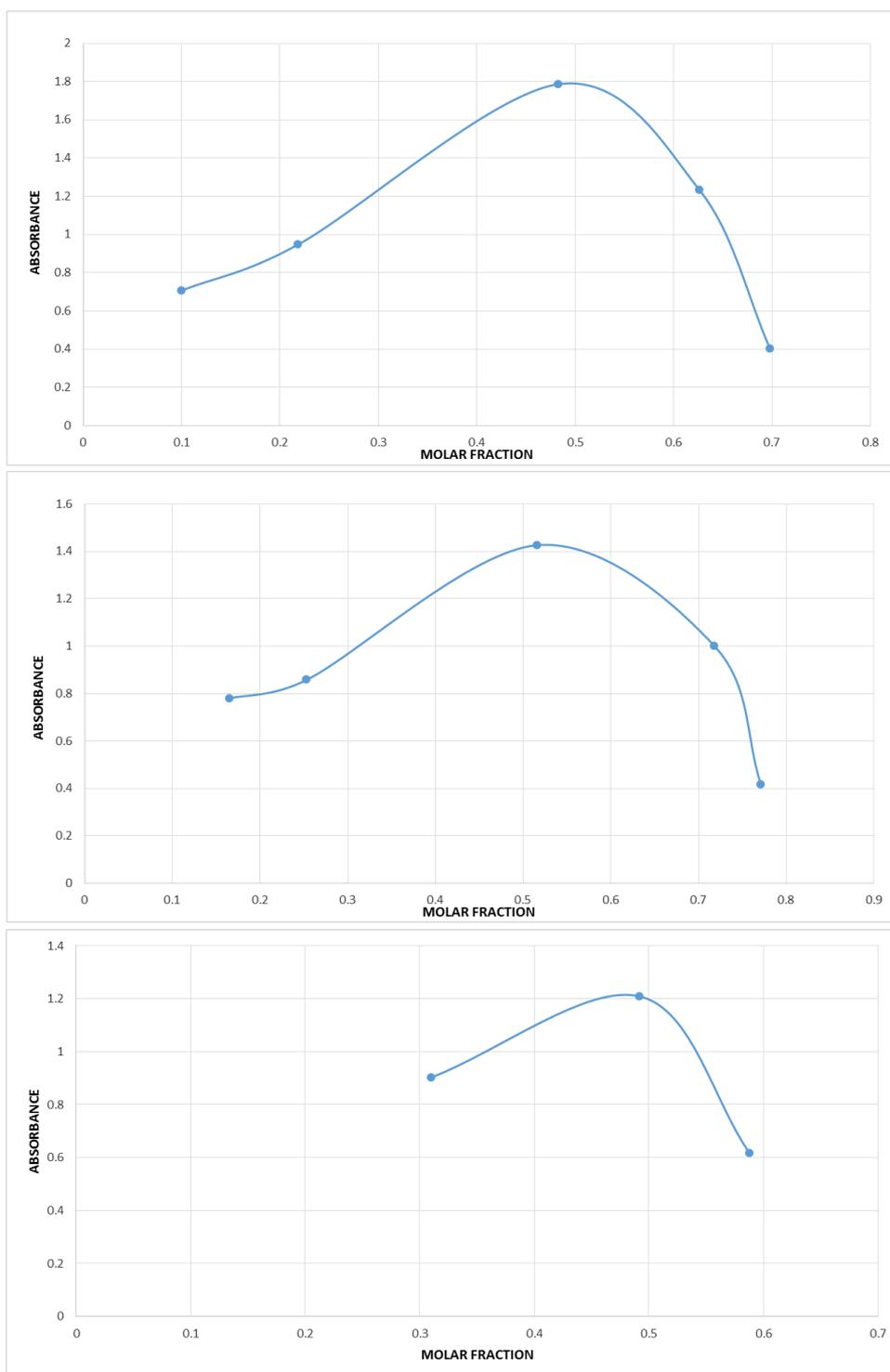


Figure S2: Job's plots of the complexes of **6** with Mg²⁺, Ca²⁺ and Ba²⁺. Molar fractions are calculated as: $X_L = C_L / (C_M + C_L)$, where C_L is the concentration of the ligand and C_M is the concentration of the metal ion.

Table S1: Relative stability^a of the tautomers, M062X in kcal/mol for **1** in acetonitrile^b.

| Level of theory | ΔE , [kcal/mol] |
|-----------------|-------------------------|
| 6-31++G** | 0.33 |
| 6-311++G** | -0.24 |
| cc-pVTZ | -0.36 |
| TZVP | -0.40 |
| def2TZVP | -0.85 |

^aPositive value, means more stable enol form and vice versa.

^bThe experimentally determined value is 0.33 kcal/mol [1].

Reference

1. Fabian, W. M. F., Antonov, L., Nedeltcheva, D., Kamounah, F. S.; Taylor, P. J.
J. Phys. Chem. A **2004**, *108*, 7603–7612.